

# Synthesis of (*E*)- and (*Z*)-Alkenylphosphonates Using Vinylboronates

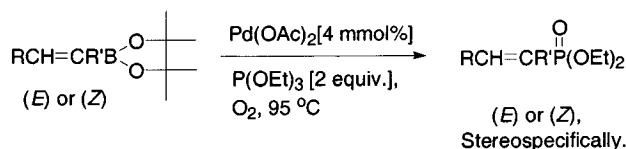
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## ABSTRACT



(*E*)- and (*Z*)-alkenylphosphonates have been prepared stereospecifically via the reaction of vinylboronate esters with triethyl phosphite in the presence of palladium acetate.

Alkenylphosphonates are routinely used to prepare biologically active molecules,<sup>1</sup> flame retardants,<sup>2</sup> polymer additives,<sup>3</sup> and other transformations<sup>4</sup> such as Michael additions,<sup>4a</sup> aminohydroxylations,<sup>4b</sup> dihydroxylations,<sup>4c</sup> aziridinations,<sup>4d,e</sup> epoxidations,<sup>4e</sup> C-glycosylations,<sup>4f</sup> Horner–Wadsworth–Emmons reactions,<sup>4g</sup> and hydrogenations.<sup>4h</sup> (*E*)- and (*Z*)-alkenylphosphonates exhibit different biological activities in

nucleotide derivatives<sup>1c,e</sup> and produce stereoisomeric intermediates in asymmetric reactions.<sup>4a–e</sup> Thus, the development of stereoselective methods to prepare (*E*)- and (*Z*)-alkenylphosphonates has become important in organic synthesis. A survey of the literature<sup>5</sup> reveals that most synthetic methods produce either (*E*)- or a mixture of (*E*)- and (*Z*)-vinylphosphonates. Coupling of dialkyl phosphites with vinyl bromides does produce (*E*)- and (*Z*)-vinylphosphonates stereoselectively,<sup>6,7</sup> but it is difficult to prepare stereochemically pure vinyl bromides. Recently, Srebnik reported a method for the synthesis of (*Z*)-vinylphosphonates.<sup>8</sup>

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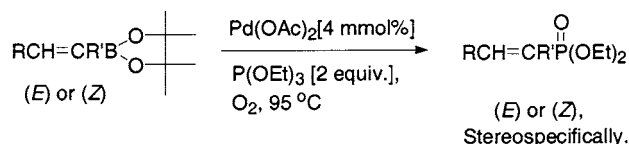
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**Table 1.** Survey of Catalysts Used in the Reaction of Pinacolato (*E*)-1-Octenylboronate with P(OEt)<sub>3</sub><sup>a</sup>

catalyst	reaction time (h)	yield <sup>b</sup> (%)
Pd(OAc) <sub>2</sub>	8	84
PdCl <sub>2</sub>	8	75
NiCl <sub>2</sub> (anhy) <sup>c</sup>	25	43
Pd(OAc) <sub>2</sub> , PPh <sub>3</sub> <sup>d</sup>	12	55
Pd(dppf)Cl <sub>2</sub>	40	incomplete reaction

<sup>a</sup> Reaction conditions: pinacolato (*E*)-1-octenylboronate (1 mmol), P(OEt)<sub>3</sub> (2 mmol), catalyst (4 mol %), 95 °C under O<sub>2</sub>. <sup>b</sup> Isolated yields. <sup>c</sup> 6 mol %. <sup>d</sup> PPh<sub>3</sub> (4 equiv) was added.

Since (*E*)- and (*Z*)-vinylboronates<sup>9,10</sup> can be prepared stereospecifically, they are ideal precursors for the stereospecific formation of (*E*)- and (*Z*)-alkenylphosphonates. We decided to explore the preparation of (*E*)- and (*Z*)-alkenylphosphonates from vinylboronates as a part of an ongoing research program focused on reactions of boronic acids and their esters.<sup>11</sup> Herein, we wish to report the results of this study (Scheme 1).

**Scheme 1**

We first examined the reaction of pinacolato (*E*)-1-octenylboronate with P(OEt)<sub>3</sub> in the presence of Pd(OAc)<sub>2</sub> in refluxing EtOH. Diethyl (*E*)-1-octenylphosphonate was formed in 54% yield in 35 h. Acetonitrile, tetrahydrofuran, and toluene were ineffective as solvents. We then examined

the reaction in the absence of solvents. When pinacolato (*E*)-1-octenylboronate was heated with triethyl phosphite and palladium acetate at 95 °C under nitrogen, diethyl (*E*)-1-octenylphosphonate was formed in 60% yield. Carrying out the reaction in the presence of oxygen improved the yield to 85%.<sup>12</sup>

A survey of various catalysts (Table 1) revealed that palladium acetate was the most effective. Addition of Ph<sub>3</sub>P to the Pd(OAc)<sub>2</sub> (or the use of Pd(dppf)Cl<sub>2</sub>) did not improve the yield. In general, an excess of P(OEt)<sub>3</sub> was required to attain the highest yields.

We then examined the reactions of various (*E*)- and (*Z*)-vinylboronates (Table 2). The Pd(OAc)<sub>2</sub>-catalyzed reaction of vinylboronates with P(OEt)<sub>3</sub> at 95 °C under an oxygen atmosphere furnished vinylphosphonates in good to high yield.<sup>13</sup> Significantly, (*E*)- and (*Z*)-vinylboronates (entries 1–6) were stereospecifically converted to corresponding (*E*)- and (*Z*)-alkenylphosphonates, respectively. The method is suitable for both aryl- and alkyl-substituted vinylboronates. Halogen functionalities (entries 3, 4, and 7) were unaffected under the reaction conditions. An  $\alpha$ -substituted vinylboronate also underwent the reaction to afford a moderate yield of expected product, although the reaction was slow (entry 8). The reaction does not appear to require a boronate ester since boronic acids also react, although an excess of P(OEt)<sub>3</sub> is required to solubilize the boronic acid. The use of P(OMe)<sub>3</sub> required a longer reaction time but produced the desired product.

In conclusion, we have developed an improved method for preparing (*E*)- and (*Z*)-alkenylphosphonates stereospecifically from vinylboronates. To the best of our knowledge, this is the first report of the conversion of vinylboronic esters and their acids to 1-alkenylphosphonates. The new method has the advantage that (*E*)- and (*Z*)-vinylphosphonates can be prepared stereospecifically in good yields using readily accessible starting materials. Further investigations are in progress.

**Table 2.** Pd(OAc)<sub>2</sub>-Catalyzed Conversion of Vinylboronates to Alkenylphosphonates,<sup>a</sup> Scheme 1

entry	R	R'	stereochemistry	time (h)	product	yield <sup>b</sup> (%)	ref <sup>c</sup>
1	Ph	H	<i>E</i>	9	PhCH=CHP(O)(OEt) <sub>2</sub> <i>E</i>	78	5i, 6
2	Ph	H	<i>Z</i>	7	PhCH=CHP(O)(OEt) <sub>2</sub> <i>Z</i>	75	5e,i, 6, 8
3	<i>p</i> -Cl-C <sub>6</sub> H <sub>4</sub>	H	<i>E</i>	9	<i>p</i> -Cl-C <sub>6</sub> H <sub>4</sub> CH=CHP(O)(OEt) <sub>2</sub> <i>E</i>	79	5i
4	<i>p</i> -Cl-C <sub>6</sub> H <sub>4</sub>	H	<i>Z</i>	6	<i>p</i> -Cl-C <sub>6</sub> H <sub>4</sub> CH=CHP(O)(OEt) <sub>2</sub> <i>Z</i>	60	5i
5	<i>n</i> -C <sub>6</sub> H <sub>13</sub>	H	<i>E</i>	8	<i>n</i> -C <sub>6</sub> H <sub>13</sub> CH=CHP(O)(OEt) <sub>2</sub> <i>E</i>	84	5b
6	<i>n</i> -C <sub>6</sub> H <sub>13</sub>	H	<i>Z</i>	6	<i>n</i> -C <sub>6</sub> H <sub>13</sub> CH=CHP(O)(OEt) <sub>2</sub> <i>Z</i>	63	13a
7	<i>p</i> -F-C <sub>6</sub> H <sub>4</sub>	H	<i>E</i>	9	<i>p</i> -F-C <sub>6</sub> H <sub>4</sub> CH=CHP(O)(OEt) <sub>2</sub> <i>E</i>	81	13b
8	Ph	CH <sub>3</sub>	<i>E</i>	120	PhCH=C(CH <sub>3</sub> )P(O)(OEt) <sub>2</sub> <i>E</i>	55	13c

<sup>a</sup> Reaction conditions: vinylboronate (1 mmol), P(OEt)<sub>3</sub> (2 mmol), and Pd(OAc)<sub>2</sub> (4 mmol %), 95 °C, O<sub>2</sub>. <sup>b</sup> Isolated yields. <sup>c</sup> Literature reference.

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**Supporting Information Available:** General experimental procedure and  $^1\text{H}$ ,  $^{13}\text{C}$ , and  $^{31}\text{P}$  NMR spectroscopic data

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